

1,3-Dipolar Cycloaddition Reactions of Nitrile-N-oxides with o-Benzoquinones[#]

Vijay Naira, K. V. Radhakrishnana, K. C. Sheela, Nigam P. Rath

a. Organic Chemistry Division, Regional Research Laboratory (CSIR), Trivandrum-695 019, Kerala, INDIA b. Department of Chemistry, University of Missouri at St. Louis, St. Louis, Mo 63121, USA.

Received 5 January 1999; revised 14 September 1999; accepted 1 October 1999

Abstract

Di- and tri-substituted o-benzoquinones on reaction with nitrile N- oxides afforded novel mono spirodioxazole derivatives. The reaction of monosubstituted o-benzoquinones with stable nitrile oxides resulted in the formation of bis adducts. © 1999 Published by Elsevier Science Ltd. All rights reserved.

Key words: o-Benzoquinones, nitrile oxide, dipolar cycloaddition, spirodioxazoles

o-Benzoquinones are unique conjugated 1,2-diones [1] that can exhibit diverse cycloaddition modes, participating either as carbodienes, heterodienes, dienophiles or heterodienophiles [2-10]. The presence of two potential dipolarophilic functionalities viz., C=C and C=O, renders o-quinones very interesting from the vantage point of dipolar cycloaddition. Although dipolar cycloaddition reactions of o-benzoquinones can conceivably lead to novel heterocyclic compounds, there has been very little work on their reaction with various dipoles. There are isolated reports on the reaction of diazomethane [11] and certain mesoionic compounds [12] with o-benzoquinones. In addition there are sketchy investigations of the dipolar cycloadditions of nitrile-N-oxides with o-bromanil as well as unsubstituted o-benzoquinone [13a], phenanthrene-9,10-quinone and 1,2-naphthoquinone [13b]. It may be pointed out that in these cases, the products have not been characterized rigorously. Against this background, and in the context of our sustained interest in the cycloaddition reactions of o-quinones, we have undertaken an investigation of their 1,3-dipolar cycloaddition with nitrile N-oxides. A preliminary report of the work has been published [14] and the details of our expanded investigations are presented here.

0040-4020/99/\$ - see front matter © 1999 Published by Elsevier Science Ltd. All rights reserved. PII: S0040-4020(99)00887-X

[&]quot;Dedicated with best regards to Dr. Lantz S. Crawley.

Results and discussion

3,5-Di-*tert*-butyl-1,2-benzoquinone 1 on treatment with 4-methyl benzohydroximoyl chloride 2a and triethylamine afforded a mixture of yellow crystalline products in 80% yield (Scheme 1).

a. $R^1 = CH_3$; $R^2 = H$; **3a**:**4a** (3:1); 80%; b. $R^1 = OCH_3$; $R^2 = H$; **3b**:**4b** (1:1); 100%; c. $R^1 = R^2 = OCH_3$; **3c**:**4c** (1:1); 99%; d. $R^1 = CI$; $R^2 = H$; **3d**:**4d** (1.6:1); 83%; e. $R^1 = H$; $R^2 = CI$; **3e**:**4e** (2:1); 100%; f. $R^1 = R^2 = H$; **3f**:**4f** (1:1); 99%

Scheme 1

The products **3a** and **4a** were isolated as yellow solids after chromatography. Their structures were established by spectral analysis, including two doublet ¹H NMR spectroscopic signals at low field typical of 3,5-di-*tert*-butyl-1,2-benzoquinone moiety and one resonance for carbonyl carbon in the ¹³C NMR spectra. The formulae were confirmed by elemental analysis and finally by single crystal X-ray analysis of **4a**[14].

The experiment was repeated with various nitrile oxides and in all cases the reaction proceeded smoothly to afford the spirodioxazoles (Scheme 1).

Analogous reaction was observed with 3-methoxy-4,6-di-*tert*-butyl-1,2-benzoquinone 5 and different benzonitrile oxides **6a-e**. In all these cases, however, an inseparable mixture of regioisomers was obtained in 1:1 ratio. The results are summarized in Scheme 2.

a. $R^1 = CH_1$, $R^2 = H$; 7a:8a (1:1); 100%; b. $R^1 = H$; $R^2 = CI$; 7b:8b (1:1); 96%; c. $R^1 = H$; $R^2 = NO_2$, 7c:8c (1:1); 97%; d. $R^1 = R^2 = OCH_3$; 7d:8d (1:1); 100%; e. $R^1 = CI$; $R^2 = H$; $R^2 = CI$; $R^3 =$

Scheme 2

The products were isolated as mixtures after column chromatography. Their structures were established by spectral analysis, the ratio of isomers (1:1) was obtained from the integration of methoxy proton signals which appeared as singlets. There was one resonance each for carbonyl carbon and C=N. Most of the other carbon atoms presented twin signals representing the two isomers. The formulae assigned were supported by satisfactory elemental analysis.

3-Methoxy-1,2-benzoquinone **9a**, 4-tert-butyl-1,2-benzoquinone **9b** and 4-methyl-1,2-benzoquinone **9c** on reaction with the nitrile oxides **6a-e** gave complex mixtures. However **9a-c** afforded isolable products with mesityl nitrile oxide **10**. For example, 3-methoxy-1,2-benzoquinone **9a**, on treatment with **10** afforded a yellow semisolid in 59% yield (Scheme 3), whose structure was assigned as **11a** by spectroscopic analysis. Similarly the structures of **11b** and **c** were established by spectral and elemental analysis.

$$R^{1}$$
 R^{2}
 R^{2}
 R^{2}
 R^{3}
 R^{4}
 R^{2}
 R^{2}
 R^{4}
 R^{4

a. $R^1 = H$; $R^2 = OCH_3$; 30 min; 59%; b. $R^1 = CMe_3$; $R^2 = H$; 15 min; 69%; c. $R^1 = CH_3$; $R^2 = H$; 30 min; 49%.

Scheme 3

The reaction of 3-methoxy-1,2-benzoquinone 9a with 2,6-dichlorobenzonitrile oxide 12 afforded a pale yellow crystalline solid in 75% yield (Scheme 4), whose structure was assigned as 13a by spectral and elemental analysis.

OMe

OMe

OMe

OMe

Ar

Benzene, RT

$$30 \text{ min. } 75\%$$

Ar

 Ar
 Ar

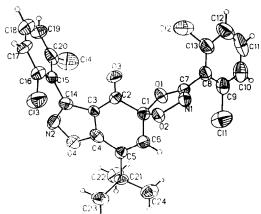
Scheme 4

The reaction of 4-tert-butyl-1,2-benzoquinone 9b and 3-methyl-1,2-benzoquinone 9c with 2,6-dichlorobenzonitrile oxide 12 afforded products resulting from addition to both C=C and C=O. For instance, treatment of 4-tert-butyl-1,2-benzoquinone 9b with 12 afforded two

products in 59% yield and these were assigned structures 14b and 15b based on spectral analysis (Scheme 5).

Scheme 5

The formulae assigned were confirmed by elemental analysis and finally by single crystal X-ray analysis of 14b.



X-ray structure of 14b

In summary, di- and tri-substituted o-benzoquinones form mixtures of mono spirodioxazoles by addition of aryl nitrile oxides to the carbonyl group. The preferential addition to the carbonyl group may be attributed to the steric hindrance imparted by the substituents on the quinone. In the case of mono substituted 1,2-benzoquinones, mesityl nitrile oxide forms bis spirodioxazoles, while 2,6-dichlorobenzonitrile oxide affords bis adducts by participation of C=C and C=O bonds. This difference in reactivity may be attributed to electronic effects of the substituents on the nitrile oxide. It is noteworthy that the mono spirodioxazole formation may be viewed as a means to obtain partially protected o-benzoquinones.

Acknowledgements: KVR and KCS thank the CSIR, New Delhi for research fellowships. The authors thank Dr. R. K. Pillai, Bristol Myers Squibb, USA and Professor G. M. Sharma,

William Paterson University, USA for high resolution NMR spectra and Dr. Jaya Prabhakaran, ICSN, CNRS, France, Dr. G. Anilkumar, University of Nijmegen, The Netherlands and Prof. D. Basavaiah, University of Hyderabad, India for elemental analysis.

Experimental Details

All reactions were carried out in oven dried glass ware (120 °C) under an atmosphere of argon. Analytical thin layer chromatography was performed on silica gel TLC plates. Purification by gravity column chromatography was carried out using silica gel (100-200 mesh). Mixtures of ethyl acetate and hexane were used as eluent. IR spectra were run on a Perkin-Elmer Model 882 and Nicolet (impact 400D FT-IR) spectrophotometers. NMR spectra were obtained using chloroform-d as solvent. Chemical shifts are given in δ scale with TMS as internal reference. Benzonitrile oxides were prepared according to the literature procedures [15-16].

Spirodioxazoles 3a and 4a

4-Methyl benzohydroximoyl chloride (0.250 g, 1.5 mmol) in benzene (10 mL) was added to a solution of 3,5-di-*tert*-butyl-1,2-benzoquinone **1** (0.220 g, 1 mmol) and triethylamine (0.152 g, 1.5 mmol) in benzene (5 mL) and stirred at room temperature for 25 min. The reaction mixture was filtered to remove triethylamine hydrochloride and the solvent was evaporated *in vacuo*. The residue was chromatographed on a silica gel column (1% and 2% ethyl acetate-hexane) afforded **3a** (0.212 g, 60%) and **4a** (0.071 g, 20%) as yellow crystalline solids (m. p. 116-118 °C and 120-122 °C respectively).

Data for 3a

IR (KBr) 2969, 1699, 1631, 1521, 1483, 1465, 1371, 1329, 1277, 1249, 1085, 1015, 936, 891, 818 cm⁻¹. ¹H NMR δ 7.48 (Ar, 4H), 6.85 (d, J = 2.4 Hz, 1H), 6.05 (d, J = 2.3 Hz, 1H), 2.40 (s, 3H), 1.25 (s, 9H), 1.15 (s, 9H). ¹³C NMR δ 192.90, 159.70, 150.45, 143.35, 142.04, 134.25, 129.27, 127.00, 121.51, 119.66, 102.53, 35.02, 34.51, 29.08, 27.98, 21.50. Anal. calcd for C₂₂H₂₇NO₃: C 74.76%, H 7.70%, N 3.96%. Found: C 74.85%, H 7.68%, N 3.85%.

Data for 4a

IR (KBr) 2979, 1685, 1631, 1485, 1473, 1351, 1189, 1119, 1092, 830, 659 cm⁻¹. ¹H NMR δ 7.51 (Ar, 4H), 6.50 (d, J = 1.7 Hz, 1H), 5.90 (d, J = 1.7 Hz, 1H), 2.40 (s, 3H), 1.30 (s, 9H), 1.15 (s, 9H). ¹³C NMR δ 193.41, 163.79, 159.76, 151.17, 142.21, 129.36, 127.03, 125.27, 119.63, 115.39, 104.29, 36.93, 35.91, 29.92, 27.89, 21.53. Anal. calcd for C₂₂H₂₇NO₃: C 74.76%, H 7.70%, N 3.96%. Found: C 74.68%, H 7.68%, N 3.92%.

Spirodioxazoles 3b and 4b

Reaction of 1 (0.220 g, 1.0 mmol) with 2b generated from 4-methoxy benzohydroximoyl chloride (0.279 g, 1.5 mmol), as described earlier yielded after

recrystallisation from hexane **3b** (0.185 g, 50%) and **4b** (0.184 g, 50%) as yellow crystalline solids (m. p. 128-130 °C and 156-158 °C respectively).

Data for 3b

IR (KBr) 2971, 2880, 1700, 1632, 1614, 1518, 1466, 1428, 1372, 1354, 1260, 1177, 1140, 1085, 936, 839 cm⁻¹. ¹H NMR δ 7.75 (Ar, 2H), 6.91 (Ar, 2H), 6.83 (d, J = 1.7 Hz, 1H), 6.03 (d, J = 1.76 Hz, 1H), 3.83 (s, 3H), 1.25 (s, 9H), 1.18 (s, 9H). ¹³C NMR δ 192.86, 162.19, 159.48, 150.33, 143.35, 134.08, 128.81, 121.60, 114. 82, 114.67, 102.40, 55.25, 35.06, 34.56, 29.13, 28.05. Anal. calcd for C₂₂H₂₇NO₄: C 71.52%, H 7.37%, N 3.79%. Found: C 71.61%, H 7.33%, N 3.82%.

Data for 4b

IR (KBr) 2978, 2880, 1688, 1630, 1611, 1520, 1463, 1350, 1311, 1258, 1182, 1123, 1093, 1029, 986, 860 cm⁻¹. 1 H NMR δ 7.78 (Ar, 2H), 6.93 (Ar, 2H), 6.47 (d, J = 1.4 Hz, 1H), 5.87 (d, J = 1.5 Hz, 1H), 3.83 (s, 3H), 1.27 (s, 9H), 1.20 (s, 9H). 13 C NMR δ 193.33, 163.50, 159.50, 151.19, 130.16, 129.67, 128.84, 125.12, 115.47, 114.76, 104.15, 55.27, 36.94, 35.92, 29.96, 27.94. Anal. calcd for $C_{22}H_{27}NO_4$: C 71.52%, H 7.37%, N 3.79%. Found: C 71.18%, H 7.21%, N 3.71%.

Spirodioxazoles 3c and 4c

Reaction of 1 (0.088 g, 0.40 mmol) with 2c generated from 3,4-dimethoxy benzohydroximoyl chloride (0.120 g, 0.60 mmol), as described earlier yielded after recrystallisation from hexane 3c (0.080 g, 50%) and 4c (0.079 g, 49%) as yellow crystals (m. p. 110-112 °C and 166-168 °C respectively).

Data for 3c

IR (KBr) 2971, 2914, 1690, 1613, 1582, 1523, 1468, 1432, 1372, 1277, 1139, 1098, 1031, 947 cm⁻¹. ¹H NMR δ 7.36 (Ar, 2H), 6.90 (Ar, 1H), 6.85 (d, J = 2.3 Hz, 1H), 6.05 (d, J = 2.3 Hz, 1H), 3.92 (s, 3H), 3.91 (s, 3H), 1.25 (s, 9H), 1.18 (s, 9H). ¹³C NMR δ 193.00, 159.70, 151.95, 150.57, 148.99, 143.40, 134.30, 121.46, 120.77, 114.92, 110.77, 109.56, 102.59, 55.97, 35.10, 34.58, 29.11, 28.04. Anal. calcd for C₂₃H₂₉NO₅: C 69.15%, H 7.32%, N 3.51%. Found: C 69.21%, H 7.34%, N 3.54%.

Data for 4c

IR (KBr) 2960, 2913, 1693, 1633, 1610, 1586, 1521, 1465, 1427, 1369, 1338, 1267, 1191, 1122, 1026, 925 cm⁻¹. 1 H NMR δ 7.36 (Ar, 2H), 6.91 (Ar, 1H), 6.50 (d, J = 1.7 Hz, 1H), 5.90 (d, J = 1.7 Hz, 1H), 3.93 (s, 3H), 3.91 (s, 3H), 1.28 (s, 9H), 1.20 (s, 9H). 13 C NMR δ 193.59, 163.85, 159.70, 152.13, 151.18, 149.07, 125.35, 120.77, 115.50, 114.86,110.80, 109.56, 104.07, 56.00, 37.00, 35.99, 30.00, 27.95. Anal. calcd for $C_{23}H_{29}NO_5$: C 69.15%, H 7.32%, N 3.51%. Found: C 69.18%, H 7.29%, N 3.52%.

Spirodioxazoles 3d and 4d

Reaction of 1 (0.220 g, 1.0 mmol) with 2d generated from 4-chloro benzohydroximoyl chloride (0.224 g, 1.5 mmol), as described earlier yielded 3d (0.230 g, 62%) and 4d (0.080 g, 21%) as yellow crystalline solids (m. p. 150-152 °C and 146-140 °C respectively).

Data for 3d

IR (KBr) 2971, 2880, 1699, 1634, 1603, 1496, 1409, 1370, 1327, 1277, 1249, 1142, 1092, 1013, 988, 937, 837 cm⁻¹. ¹H NMR δ 7.59 (Ar, 4H), 6.85 (d, J = 2.3 Hz, 1H), 6.04 (d, J = 2.3 Hz, 1H), 1.25 (s, 9H), 1.18 (s, 9H). ¹³C NMR δ 193.00, 159.70, 150.95, 143.40, 137.82, 134.48, 129.03, 128.40, 121.20, 121.11, 102.00, 35.19, 34.61, 29.14, 28.04. Anal. calcd for C₂₁H₂₄NO₃Cl: C 67.46%, H 6.47%, N 3.75%. Found: C 67.51%, H 6.49%, N 3.78%.

Data for 4d

IR (KBr) 2962, 2874, 1694, 1640, 1600, 1485, 1357, 1270, 1196, 1101, 1000, 838, 663 cm⁻¹.
¹H NMR δ 7.61 (Ar, 4H), 6.50 (d, J = 1.7 Hz, 1H), 5.90 (d, J = 1.7 Hz, 1H), 1.27 (s, 9H), 1.20 (s, 9H).
¹³C NMR δ 193.00, 164.74, 159.79, 151.58, 138.71, 129.84, 129.12, 126.30, 121.72, 116.16, 104.44, 37.69, 36.74, 30.66, 28.65. Anal. calcd for C₂₁H₂₄NO₃Cl: C 67.46%, H 6.47, N 3.75. Found: C 67.22%, H 6.55%, N 3.52%.

Spirodioxazoles 3e and 4e

Reaction of 1 (0.220 g, 1.0 mmol) with **2e** generated from 3-chloro benzohydroximoyl chloride (0.284 g, 1.5 mmol), as described earlier yielded **3e** (0.235 g, 63%) and **4e** (0.139 g, 37%) as yellow crystalline solids (m. p.125-127°C and 129-131°C respectively).

Data for 3e

IR (KBr) 2967, 2877, 1696, 1657, 1575, 1487, 1369, 1327, 1279, 1251, 1140, 1114, 1072, 941, 892 cm⁻¹. ¹H NMR δ 7.58 (Ar, 4H), 6.86 (d, J = 2.3 Hz, 1H), 6.04 (d, J = 2.3 Hz, 1H), 1.25 (s, 9H), 1.18 (s, 9H). ¹³C NMR δ 193.00, 159.60, 151.03, 143.40, 134.79, 134.50, 131.68, 130.01, 127.10, 125.20, 124.28, 121.11, 102.05, 35.19, 34.61, 29.14, 28.04. Anal. calcd for C₂₁H₂₄NO₃Cl: C 67.46%, H 6.47%, N 3.75%. Found: C 67.55%, H 6.50%, N 3.72%.

Data for 4e

IR (KBr) 2970, 2878, 1695, 1644, 1577, 1485, 1354, 1267, 1192, 1120, 1075, 997, 922, 811 cm⁻¹. ¹H NMR δ 7.60 (Ar, 4II), 6.51 (d, J = 1.7 Hz, 1H), 5.90 (d, J = 1.7 Hz, 1H), 1.27 (s, 9H), 1.20 (s, 9H). ¹³C NMR δ 193.16, 164.08, 158.81, 150.83, 134.88, 131.82, 130.10, 127.10, 125.69, 125.23, 124.22, 115.44, 104.44, 37.03, 36.08, 30.00, 27.95. Anal. calcd for C₂₁H₂₄NO₃Cl: C 67.46%, H 6.47%, N 3.75%. Found: C 67.55%, H 6.44%, N 3.72%.

Spirodioxazoles 3f and 4f

Reaction of 1 (0.100 g, 0.45 mmol) with 2f generated from benzohydroximoyl chloride (0.079 g, 0.50 mmol), as described earlier yielded 3f (0.083 g, 49%) and 4f (0.085 g, 50%) as yellow crystalline solids (m. p. 144-146 °C and 120-122 °C respectively).

Data for 3f

IR (KBr) 2971, 2877, 1700, 1634, 1511, 1451, 1362, 1325, 1277, 1249, 1140, 1090, 1014, 938, 890 cm⁻¹. ¹H NMR δ 7.83 (Ar, 2H), 7.46 (Ar, 3H), 6.85 (d, J = 2.3 Hz, 1H), 6.05 (d, J = 2.3 Hz, 1H), 1.25 (s, 9H), 1.18 (s, 9H). ¹³C NMR δ 193.00, 159.80, 151.60, 144.01, 134.45, 131.65, 128.66, 127.16, 122.30, 121.46, 102.50, 35.89, 35.60, 29.14, 28.07. Mass (EIMS): 339 (M⁺), 309 (M-NO), 283, 222, 207, 149, 119, 105, 91, 57. Anal. calcd for C₂₁H₂₅NO₃: C 74.31%, H 7.42%, N 4.13%. Found: C 74.38%, H 7.40%, N 4.15%.

Data for 4f

IR (KBr) 2976, 2880, 1688, 1630, 1583, 1475, 1454, 1349, 1249, 1184, 1120, 1092, 988, 922, 816 cm⁻¹. ¹H NMR δ 7.86 (Ar, 2H), 7.48 (Ar, 3H), 6.50 (d, J = 1.7 Hz, 1H), 5.90 (d, J = 1.7 Hz, 1H), 1.28 (s, 9H), 1.20 (s, 9H). ¹³C NMR δ 193.00, 164.50, 159.90, 151.50, 131.77, 128.71, 127.13, 125.40, 122.10, 115.47, 104.60, 37.10, 36.20, 29.97, 27.95. Mass (EIMS): 339 (M⁺), 309 (M-NO), 283, 222, 207, 164, 149, 119, 105, 91, 57. Anal. calcd for C₂₁H₂₅NO₃: C 74.31%, H 7.42%, N 4.13%. Found: C 74.35%, H 7.39%, N 4.18%.

Spirodioxazoles 7a and 8a

Reaction of 3-methoxy-4,6-di-*tert*-butyl-1,2-benzoquinone **5** (0.100 g, 0.40 mmol) with **6a** generated from 4-methyl benzohydroximoyl chloride (0.135 g, 0.80 mmol), as described earlier afforded inseparable mixture of isomers **7a** and **8a** (0.171 g, 100%) as pale yellow solid. IR (film) 2962, 2874, 1694, 1634, 1573, 1472, 1351, 1276, 1188, 1087, 952, 878, 824 cm⁻¹. ¹H NMR δ 7.49 (Ar, 8H), 6.98 (s, 1H), 6.40 (s, 1H), 3.75 (s, 3H), 3.72 (s, 3H), 2.39 (s, 3H), 2.29 (s, 3H), 1.29 (s, 18H), 1.23 (s, 18H). ¹³C NMR δ 191.29, 190.09, 159.94, 159.77, 149.11, 144.67, 142.17, 138.38, 129.38, 129.35, 127.09, 127.04, 119.57, 119.41, 105.91, 103.03, 60.00, 58.44, 36.62, 35.55, 34.86, 34.18, 29.81, 29.66, 29.22, 28.95, 21.55, 21.43. Anal. calcd for C₂₃H₂₉NO₄: C 72.04%, H 7.62%, N 3.65%. Found: C 72.18%, H 7.45%, N 3.49%.

Spirodioxazoles 7b and 8b

Reaction of 3-methoxy-4,6-di-*tert*-butyl-1,2-benzoquinone **5** (0.09 g, 0.36 mmol) with **6b** generated from 3-chloro benzohydroximoyl chloride (0.100 g, 0.72 mmol), as described earlier afforded inseparable mixture of isomers **7b** and **8b** (0.171 g, 100%) as pale yellow solid. IR (film) 2960,1697, 1637, 1571, 1483, 1439, 1357, 1269, 1077, 1033, 956, 884 cm⁻¹. ¹H NMR δ 7.60 (Ar, 8H), 7.02 (s, 1H), 6.45 (s, 1H), 3.79 (s, 3H), 3.72 (s, 3H), 1.31 (s, 18H), 1.24 (s, 18H). ¹³C NMR δ 191.02, 189.89, 158.86, 158.68, 144.69, 144.18, 138.69, 134.73, 132.01, 131.71, 129.98, 127.98, 126.88, 126.40, 125.03, 123.93, 123.78, 107.50, 106.00, 60.29, 58.47, 36.51, 35.52, 34.93, 34.21, 29.68, 29.47, 29.05, 28.90. Anal. calcd for C₂₂H₂₆NO₄Cl: C 65.42%, H 6.49%, N 3.47%. Found: C 65.26%, H 6.67%, N 3.35%.

Spirodioxazoles 7c and 8c

Reaction of 3-methoxy-4,6-di-*tert*-butyl-1,2-benzoquinone **5** (0.100 g, 0.40 mmol) with **6c** generated from 3-nitro benzohydroximoyl chloride (0.160 g, 0.80 mmol), as described earlier afforded inseparable mixture of **7c** and **8c** (0.161 g, 97%) as yellow solid. IR (film) 2962, 1694, 1634, 1546, 1479, 1351, 1270, 1128, 1034, 953, 879, 744 cm⁻¹. ¹H NMR δ 8.70 (s, 2H), 8.39 (d, 2H), 8.21 (d, 2H), 7.71 (m, 2H), 7.04 (s, 1H), 6.48 (s, 1H), 3.78 (s, 3H), 3.73 (s, 3H), 1.31 (s, 18H), 1.25 (s, 18H). ¹³C NMR δ 191.02, 189.87, 158.51, 158.29, 148.46, 145.50, 145.01, 144.05, 139.09, 132.59, 130.17, 130.12, 126.97, 126.29, 122.02, 106.85, 105.50, 60.76, 58.74, 36.72, 35.51, 34.89, 34.42, 29.89, 29.64, 29.40, 29.23. Anal. calcd for $C_{22}H_{26}N_2O_6$: C 63.76%, H 6.32%, N 6.76%. Found: C 63.38%, H 6.53%, N 6.58%.

Spirodioxazoles 7d and 8d

Reaction of 3-methoxy-4,6-di-*tert*-butyl-1,2-benzoquinone **5** (0.100 g, 0.40 mmol) with **6d** generated from 3,4-dimethoxy benzohydroximoyl chloride (0.170 g, 0.80 mmol), as described earlier afforded inseparable mixture of regioisomers **7d** and **8d** (0.171 g, 100%) as pale yellow solid. IR (KBr) 2969, 1694, 1640, 1607, 1510, 1472, 1378, 1344, 1270, 1148, 1034, 960, 879 cm⁻¹. ¹H NMR δ 7.26 (Ar, 6H), 7.03 (s, 1H), 6.42 (s, 1H), 3.95 (s, 6H), 3.80 (s, 3H), 3.74 (s, 3H), 1.31 (s, 18H), 1.25 (s, 18H). ¹³C NMR δ 191.04, 189.86, 160.05, 159.80, 152.12, 149.02, 148.87, 144.78, 144.63, 144.54, 129.95, 126.07, 120.73, 114.62, 110.77, 109.43, 60.08, 58.50, 55.87, 36.57, 35.52, 34.90, 34.18, 29.74, 29.56, 29.11, 28.99. Anal. calcd for C₂₄H₃₁NO₆: C 67.11%, H 7.27%, N 3.26%. Found: C 66.98%, H 7.46%, N 3.10%.

Spirodioxazoles 7e and 8e

Reaction of 3-methoxy-4,6-di-*tert*-butyl-1,2-benzoquinone **5** (0.135 g, 0.54 mmol) with **6e** generated from 4-Chloro benzohydroximoyl chloride (0.208 g, 1.08 mmol), as described earlier afforded inseparable mixture of isomers **7e** and **8e** (0.212 g, 98%) as yellow solid. IR (film) 2969, 2874, 1694, 1634, 1607, 1566, 1492, 1364, 1310, 1270, 1094, 1061, 1020, 946, 838 cm⁻¹. ¹H NMR δ 7.65 (Ar, 8H), 7.02 (s, 1H), 6.41 (s, 1H), 3.75 (s, 3H), 3.68 (s, 3H), 1.29 (s, 18H), 1.25 (s, 18H). ¹³C NMR δ 191.20, 189.98, 159.24, 158.84, 150.41, 145.68, 138.81, 138.66, 137.92, 131.98, 128.97, 128.31, 126.30, 121.72, 120.58, 107.40, 106.82, 60.29, 58.50, 36.44, 35.60, 34.96, 34.24, 29.64, 29.50, 29.03, 28.93. Anal. calcd for C₂₂H₂₆NO₄Cl: C 65.42%, H 6.49%, N 3.47%. Found: C 65.29%, H 6.72%, N 3.42%.

Bis-spirodioxazole 11a

To a solution of 3-methoxy catechol (0.140 g, 1.00 mmol) and mesityl nitrile oxide 10 (0.405 g, 2.5 mmol) in benzene (10 mL), silver carbonate (0.690 g, 2.5 mmol) was added and stirred at room temperature for 30 min. The inorganic material was removed by filtering through a short celite column. The filtrate was then concentrated *in vacuo* and the residue was charged on a silica gel column. Elution with 30% ethyl acetate in hexane afforded 11a (0.270g,

59%) as a yellow semisolid. IR (film) 2968, 2917, 2866, 1667, 1616, 1452, 1327, 1248, 1073, 1022, 903, 852 cm⁻¹. ¹H NMR δ 6.92 (Ar, 4H), 5.62 (m. 1H), 5.35 (brs, 1H), 4.88 (m, 1H), 3.88 (s, 3H), 2.27 (m, 18H). ¹³C NMR δ 158.95, 158.04, 143.29, 141.23, 140.96, 140.75, 138.99, 138.81, 129.77, 128.31, 116.85, 105.80, 104.06, 55.45, 20.97, 19.53, 19.35. Anal. calcd for C₂₇H₂₈N₂O₅: C 70.42%, H 6.13%, N 6.08%. Found: C 70.64%, H 6.55%, N 5.98%.

Bis-spirodioxazole 11b

A solution of 4-*tert*-butyl-1,2-benzoquinone (0.164 g, 1 mmol) and mesityl nitrile oxide **10** (0.405 g, 2.5 mmol) in benzene (15 mL) was stirred at room temperature for 15 min. The reaction mixture was concentrated *in vacuo* and the residue on silica gel column chromatography (5% ethyl acetate in hexane) afforded **11b** (0.335 g, 69%) as a pale yellow semisolid. IR (film) 2969, 2928, 2874, 1635, 1613, 1452, 1378, 1324, 1263, 1081, 1007, 919, 858 cm⁻¹. ¹H NMR δ 6.86 (Ar, 4H), 6.40 (m, 1H), 6.11 (m, 1H), 5.80 (brs, 1H), 2.24 (m, 18H), 1.16 (s, 9H). ¹³C NMR δ 158.50, 157.90, 141.37, 140.67, 138.96, 138.71, 130.61, 128.62, 128.49, 126.80, 124.38, 124.01, 118.19, 110.81, 109.51, 34.51, 27.98, 21.21, 19.90, 19.71. Anal. calcd for C₃₀H₃₄N₂O₄: C 74.05%, H 7.04%, N 5.76%. Found: C 73.89%, H 7.31%, N 5.46%.

Bis-spirodioxazole 11c

Reaction of 4-methyl catechol **9c** (0.155 g, 1.25 mmol) and mesityl nitrile oxide **10,** as described earlier for **11a** afforded **11c** (0.273 g, 49%) as colorless solid. IR (film) 2969, 2928, 2854, 1620, 1445, 1324, 1216, 1175, 1142, 1074, 1020, 912, 858 cm⁻¹. ¹H NMR δ 6.71 (Ar, 4H), 5.85 (brs, 1H), 5.21 (brs, 1H), 4.83 (m, 1H), 2.10 (m, 21H). ¹³C NMR δ 158.91, 158.29, 141.55, 140.33, 139.59, 139.50, 130.18, 129.14, 125.21, 122.68, 118.31, 109.80, 109.39, 22.30, 21.95, 21.87, 20.46. Anal. calcd for C₂₇H₂₈N₂O₄: C 72.95%, H 6.35%, N 6.30%. Found: C 72.49%, H 6.91%, N 6.59%.

Bisadduct 13a

Reaction of 3-methoxy catechol **9a** (0.210 g, 1.5 mmol) and 2,6-dichloro benzonitrile oxide **12** (0.705 g, 3.75mmol) as described earlier afforded **13a** (0:574 g, 75%) as a yellow solid. (m. p. 203-205 °C). IR (KBr) 3063, 2942, 2854, 1721, 1640, 1566, 1431, 1303, 1263, 1209, 912, 784 cm⁻¹. ¹H NMR δ 7.29 (m, 6H), 5.91 (dd, J = 3.5 Hz, 8.89 Hz, 1H), 5.82 (d, J = 3.4 Hz, 1H), 4.97 (d, J = 8.9 Hz, 1H), 3.78 (s, 3H). ¹³C NMR δ 180.00, 154.95, 151.68, 151.20, 136.38, 135.34, 135.10, 132.97, 131.48, 128.46, 128.19, 126.75, 121.00, 110.55, 107.11, 55.89, 55.59. Anal. calcd for C₂₁H₉N₂O₅Cl₄: C 49.15%, H 2.16%, N 5.46%, Cl 27.64%. Found: C 48.46%, H 2.63%, N 5.25%, Cl 28.07%.

Bis adducts 14b and 15b

Reaction of 4-*tert*-butyl-1,2-benzoquinone **9b** (0.164 g, 1.00 mmol) and 2,6-dichloro benzonitrile oxide **12**, (0.470 g, 2.5 mmol) as described earlier for **11b** afforded **14b** (0.226 g, 42%) and **15b** (0.091 g, 17%) as colorless solids.

Specral data for 14b

IR (film) 2969, 2915, 2874, 1708, 1661, 1620, 1566, 1438, 1317, 1270, 1202, 1121, 885, 791,730 cm $^{-1}$. 1 H NMR δ 7.24 (m, 6H), 6.49 (brs, 1H), 6.01 (d, 1H), 5.06 (d, 1H), 1.39 (s, 9H). 13 C NMR δ 184.53, 165.80, 154.90, 151.75, 136.35, 132.83, 131.31, 128.45, 128.20, 126.27, 121.24, 106.67, 79.17, 55.28, 37.20, 30.06. Anal. calcd for $C_{24}H_{18}N_{2}O_{4}Cl_{4}$: C 53.36%, H 3.36%, N 5.19%, Cl 26.25%. Found: C 53.01%, H 3.78%, N 5.01%, Cl 25.97%.

Spectral data for 15b

IR (KBr) 2975, 1721, 1573, 1458, 1445, 1317, 1202, 1135, 993, 946, 919, 879, 791, 737 cm 1 . 1 H NMR 7.41 (Ar, 6H), 6.43 (s, 1H), 1.47 (s, 9H). 13 C NMR 182.85, 174.70, 155.00, 154.27, 142.20, 136.66, 133.06, 129.01, 128.36, 128.20, 125.41, 121.27, 103.95, 35.15, 28.73. Anal. calcd for $C_{24}H_{16}N_2O_4$: C 53.56%, H 3.00%, N 5.20%, Cl 26.35%. Found: C 53.45%, H 3.18%, N 5.08%, Cl 26.05%.

<u>Crystal data for 14b</u> $C_{24}H_{16}N_2O_4Cl_4$, colorless crystalline solid, 0.28x0.20x0.14 mm., monoclinic, space group $P2_1$, Unit cell dimensions: a = 12.5823(2) Å. alpha = 90 °, b = 17.5486(2) Å, beat = 95.635(1)°, c = 12.9804(2) Å, gamma = 90 °. R indices (all data) R1 = 0.1364, WR2 = 0.1960, Volume Z = 2852.24(7) Å 3 , 2. Density (Calculated) = 1.253 Mg/m 3 . F(000) = 1096; Absorption coefficient = 0.444 mm $^{-1}$.

Bis adducts 14c and 15c

Reaction of 3-methyl catechol **9c** (0.223 g, 1.8 mmol) and 2,6-dichloro benzonitrile oxide **12**, (1.00 g, 5.31 mmol) as described earlier for **13a** afforded **14c** (0.576 g, 65%) and **15c** (0.133 g, 15%) as colorless solids.

Spectral data for 14c

IR (KBr) 3076, 2950, 2820, 1735, 1654, 1560, 1431, 1303, 1202, 1128, 1088, 939, 892, 784 cm⁻¹. 1 H NMR δ 7.22 (m, 6H), 6.31 (s, 1H), 5.63 (d, J = 9.0 Hz, 1H), 5.00 (d, J = 9.3 Hz, 1H), 2.25 (s, 3H). 13 C NMR δ 183.17, 155.81, 154.85, 151.79, 136.37, 132.84, 131.39, 128.19, 127.70, 121.20, 106.48, 80.45, 55.59, 21.01. Anal. calcd for $C_{21}H_{12}N_2O_4Cl_4$: C 50.63%, H 2.43%, N 5.62%, Cl 28.47%. Found: C 50.98%, H 2.74%, N 5.53%, Cl 28.20%.

Spectral data for 15c

IR (KBr) 3083, 2955, 2921, 2854, 1728, 1600, 1458, 1452, 1384, 1330, 1209, 1148, 1094, 993, 784 cm⁻¹. 1 H NMR 7.44 (Ar, 6H), 6.46 (s, 1H), 2.39 (s, 3H). 13 C NMR 182.70, 174.79, 155.93, 154.24, 136.62, 131.99, 128.33, 128.26, 128.21,125.26,121.22, 103.96, 15.55. Anal. calcd for $C_{21}H_{10}N_2O_4Cl_4$: C 50.84%, H 2.03%, N 5.65%, Cl 28.58%. Found: C 51.02%, H 2.38%, N 5.45%, Cl 28.32%.

References

- 1. Patai, S.; Rappoport, Z. In "The Chemistry of Quinonoid Compounds". Vol. 2; John Wiley and Sons, New York, 1988.
- 2. Friedrichsen, W.; Betz, M.; Buldt, E.; Jurgens, H. J.; Smidt, R.; Schwarz, I.; Visser, K.; Ann. Chem. 1978, 440
- 3. Ansell, M. F.; Gosden, A. F.; Leslie, V. J.; Murray, R. A. J. Chem. Soc. (C), 1971; 1401.
- 4. Nair, V.; Kumar, S. Synlett, 1996; 1143.
- 5. Nair, V.; Kumar, S. J. Chem. Soc., Perkin Trans. I. 1996, 443.
- 6. Nair, V.; Kumar, S. J. Chem. Soc., Chem. Commun. 1994, 1341.
- 7. Nair, V.; Kumar, S.; Rath, N. P.; Morton, G. O. Chem. Lett. 1995, 383.
- 8. Nair, V.; Kumar, S.; Williard, P. G. Tetrahedron, 1995; 35, 1605.
- 9. Nair, V.; Kumar, S. Synth. Commun. 1996, 26, 697.
- 10. Nair, V.; Kumar, S. Tetrahedron, 1996, 52, 4029.
- 11. Komissarova, N. L.; Belostotskaya, I. S.; Vol'eva, V. B.; Dzhuraya, E. V.; Novikova, I. A.; Ershov, V. V.; *Izv. Akad. Nauk. SSSR, Ser. Khim. (Eng. Transl.).* **1981**, 2360.
- 12. a. Friedrichsen, W.; Schwarz, I. Tetrahedron Lett. 1977, 3581
 - b. Friedrichsen, W.; Schroer, W. D.; Debaerdemaeker, T. Ann. Chem., 1980; 1836.
 - c. Friedrichsen, W.; Shmidt, R.; van Hummel, G. J.; van den Ham, D. H. W. Ann., 1981; 521.
 - d. Friedrichsen, W.; Kappe, T.; Bottcher, A. Heterocycles, 1982; 19: 1083.
 - e. Friedrichsen, W.; Krueger, C.; Kujath, E.; Liebezeit, G.; Mohr, S. Tetrahedron Lett., 1979; 237.
 - f. Friedrichsen, W.; Schroer, W. D. Ann. Chem., 1980; 1850.
- a. Awad, W. I.; Omran, S. M. A. R.; Sobhy, M. J. Org. Chem, 1966; 31: 331
 b. Awad, W. I.; Sobhy, M. Can. J. Chem, 1969; 47: 1473.
- 14. Nair, V.; Radhakrishnan, K. V.; Nair, A. G.; Bhadbhade, M. M. Tetrahedron Lett, 1996; 37: 5623
- 15 Beltrame, P.; Dandoni, A.; Barbaro, G.; Gelli, G.; Loi, A.; Steffe, S. J Chem. Soc., Perkin Trans. II, 1978; 607.
- 16. Grundmann, C.; Dean, J. M. J. Org. Chem., 1965; 30: 2809.